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## Inhibitory effect of $\text{NaN}_3$ on the $\text{F}_0\text{F}_1$ ATPase of submitochondrial particles as related to nucleotide binding

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The inhibitory effects of  $\text{NaN}_3$  on the  $\text{F}_0\text{F}_1$  ATPase of beef heart submitochondrial particles were investigated. It was shown that  $\text{NaN}_3$  inhibited the ATPase activity only in the presence of ATP or ADP and the inhibition proceeded slowly. Analysis of the time-course of the inhibition process lead to a conclusion that an ATP binding site which has an apparent  $K_d$  of  $14.0 \pm 8.7 \mu\text{M}$  is responsible for the increase of  $\text{NaN}_3$  sensitivity. This value agreed well with the low  $K_m$  of ATP hydrolysis characterized before (Muneyuki, E., and Hirata, H. (1988) FEBS Lett. 234, 455–458) and in the range of so-called bi-site catalysis. The same conclusion was derived as for isolated  $\text{F}_1$  ATPase. From similar analysis, the  $K_d$  of this site for ADP was deduced to be  $1.34 \pm 0.45 \mu\text{M}$ , which also agreed with that reported by Pedersen (Pedersen, P.L. (1975) Biochem. Biophys. Res. Commun. 64, 610–616) and also in the same range as reported for the low  $K_m$  of ATP synthesis by activated submitochondrial particles. These results suggest that hydrolysis through the low  $K_m$  mode of ATPase reaction leads the enzyme  $\text{NaN}_3$  sensitive form and this reaction cycle corresponds to the low  $K_m$  mode of ATP synthesis.

### Introduction

Membrane-bound proton translocating ATPase ( $\text{F}_0\text{F}_1$  ATPase) has a central role in energy transduction by converting the protonic electrochemical potential to the chemical potential of ATP [1]. In these 30 years, the mechanism of ATP hydrolysis has been intensively studied using  $\text{F}_1$  portion and many details have become known [2–6]. One of the most prominent characteristics of this enzyme is the apparent negative cooperativity of ATP hydrolysis [7]. Under steady-state conditions, non-linear fitting analyses revealed two apparent  $K_m$  values of 1–10  $\mu\text{M}$  and 100–300  $\mu\text{M}$  [1,8–10]. Sometimes, a third  $K_m$  of around or above 1 mM has been reported [1,8,10]. The  $V_{\max}$  corresponding to the  $K_m$  of 1–10  $\mu\text{M}$  is usually in the range of 1–10  $\mu\text{mol mg}^{-1} \text{min}^{-1}$  and the  $V_{\max}$  corresponding to

higher  $K_m$ 's are about ten fold higher than the former. In addition, under a single-turnover conditions, a catalytic site with very high affinity ( $K_d = 10^{-12} \text{ M}$ ) and very low turnover rate ( $10^{-4} \text{ s}^{-1}$ ) has been reported for mitochondrial  $\text{F}_1$  ATPase [11,12,3]. The features of this single turnover are well consistent with the binding change mechanism proposed by Boyer's group [2,9,13] and these reactions with different  $K_d$  or  $K_m$  values are called uni-site, bi-site and tri-site catalysis, respectively [14]. Previously, we reported that both the ATPase reaction and proton translocation exhibited negative cooperativity characterized with two apparent  $K_m$  values of about 10  $\mu\text{M}$  and 200  $\mu\text{M}$  using submitochondrial particles (SMP) [15]. However, characteristics of these kinetic modes with different  $K_m$  values were not clear. In order to get insights into the nature of these two kinetic modes, it might be useful to characterize inhibitor sensitivity. Among many inhibitors which act to  $\text{F}_0\text{F}_1$  ATPase, Vasilyeva et al. reported that  $\text{NaN}_3$  acts on the ATPase of AS particles only in the presence of ATP [16]. On the other hand, Noumi et al. reported that uni-site catalysis of  $\text{F}_1$  ATPase from *E. coli* was not inhibited by  $\text{NaN}_3$  [17]. These results mean that ATP binding to some catalytic or non-catalytic site other than the uni-site of the  $\text{F}_0\text{F}_1$

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Abbreviations:  $\text{EF}_1$ ,  $\text{F}_1$ -ATPase from *Escherichia coli*; SMP, submitochondrial particles.

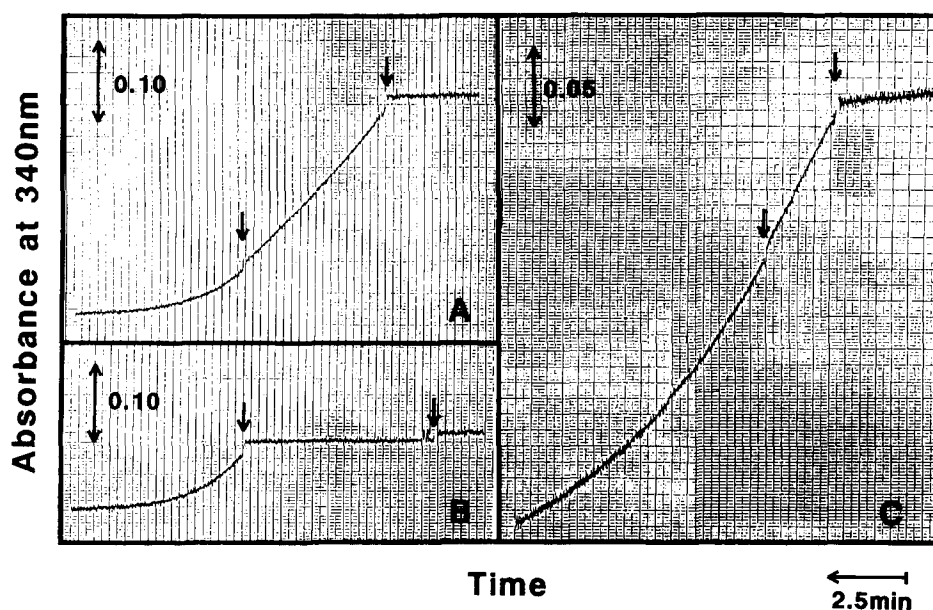


Fig. 1. Effect of  $\text{NaN}_3$  on ATPase activity of submitochondrial particles. ATPase activity was measured spectrophotometrically as described in Materials and Methods. (A) The reaction was started by the addition of ATP ( $400 \mu\text{M}$ ) as indicated by the right arrow and  $\text{NaN}_3$  ( $500 \mu\text{M}$ ) was added subsequently as indicated by the left arrow. (B) The same as in (A), except the order of addition was reversed.  $\text{NaN}_3$  ( $500 \mu\text{M}$ ) was added first at the right arrow and ATP ( $400 \mu\text{M}$ ) was added subsequently as indicated by the left arrow. (C) The same as in (A), except the ATP concentration was lowered to  $8 \mu\text{M}$ . The amount of SMP was increased so that the activity was measured more precisely.

ATPase greatly enhances  $\text{NaN}_3$  sensitivity. Here we analyzed the inhibitory effect of  $\text{NaN}_3$  as a function of ATP concentration and the results indicated that binding of ATP to a site which has an apparent  $K_d$  of  $14.0 \pm 8.7 \mu\text{M}$  increases  $\text{NaN}_3$  sensitivity. \* The apparent  $K_d$  for ATP is close to the low  $K_m$  characterized before [15] corresponding to that of so-called bi-site catalysis. The same conclusion was derived for isolated  $F_1$  ATPase. It was also shown that  $\text{NaN}_3$  sensitivity was increased by binding of ADP and an apparent  $K_d$  for ADP was deduced to be  $1.34 \pm 0.45 \mu\text{M}$  and partially reduced by inorganic phosphate.

## Materials and Methods

### Reagents

Pyruvate kinase was purchased from Sigma (type II, P-1506). Lactate dehydrogenase was purchased from Boehringer (127 876). Other reagents were of the highest grade commercially available.

\* In the present study, it is assumed that the enzyme fraction which binds ATP under steady-state hydrolysis conditions is sensitive to  $\text{NaN}_3$ . On this assumption, the  $K_d$  obtained in this study should not be regarded as the real  $K_d$  which reflects the first binding step of enzyme reaction and it should be rather close to the apparent  $K_m$  of ATPase reaction. However, for simplicity, we use the term ' $K_d$ ' or 'apparent affinity' throughout this article.

### Preparation of submitochondrial particles and $F_1$ ATPase

SMP were prepared from beef heart mitochondria suspended in  $0.25 \text{ M}$  sucrose,  $10 \text{ mM}$  Tris-HCl (pH 7.5),  $5 \text{ mM}$   $\text{MgSO}_4$ , by sonication followed by centrifugation [18]. Typically, specific activity for ATP hydrolysis at  $4 \text{ mM}$  of ATP was between  $2$  and  $3 \mu\text{mol mg}^{-1} \text{ min}^{-1}$  (at  $25^\circ\text{C}$ ) and oligomycin sensitivity was more than  $95\%$ .  $F_1$  ATPase was isolated by the chloroform treatment method by Linnet et al. [19] and further purified by DEAE-Toyopearl ion-exchange chromatography. The specific activity of the purified enzyme was between  $50$  and  $60 \mu\text{mol mg}^{-1} \text{ min}^{-1}$ .

### Measurement of ATPase activity

ATPase activity was measured spectrophotometrically using ATP regenerating system [20]. The assay mixture contained  $50 \text{ mM}$  Tris- $\text{SO}_4$  (pH 8.0),  $2 \text{ mM}$   $\text{MgSO}_4$ ,  $7 \text{ mM}$  KCN,  $2.5$ – $5 \text{ mM}$  phosphoenolpyruvate,  $0.3 \mu\text{g/ml}$  FCCP,  $34 \mu\text{g/ml}$  pyruvate kinase,  $45 \mu\text{g/ml}$  lactate dehydrogenase,  $0.2 \text{ mM}$  NADH, and indicated amount of ATP and  $\text{NaN}_3$ . As for  $F_1$  ATPase, FCCP and KCN were omitted. Coupling enzymes were used without desalting.

When necessary, ATP solution was pre-incubated with the assay mixture and contaminating ADP was completely converted to ATP before assay.

Rate constants ( $\tau$ ) of inhibition process in the presence of ATP or ADP were deduced by measuring the

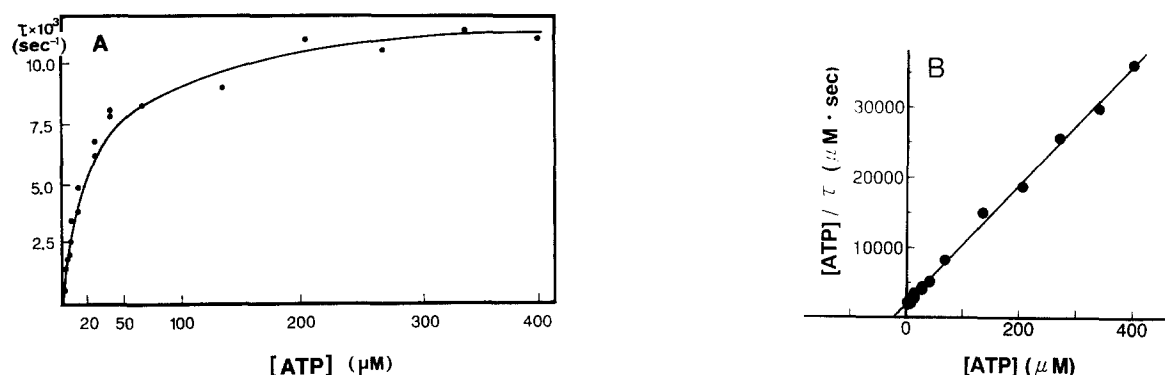


Fig. 2. ATP concentration dependency of the rate constant of inhibition process by  $\text{NaN}_3$ . (A) Rate constants of  $\text{NaN}_3$  inhibition were plotted against ATP concentrations. At ATP concentrations higher than  $67.8 \mu\text{M}$ , the rate constants were derived by fitting the time-course of inhibition as shown in Fig. 1B with an exponential curve. At lower ATP concentrations, the reaction was started by the addition of ATP at the indicated concentrations in the presence of  $\text{NaN}_3$ . After a period, additional ATP was added to  $400 \mu\text{M}$  and the remaining activity was compared to the activity at  $400 \mu\text{M}$  of ATP without  $\text{NaN}_3$ . The rate constants were derived from the percentage of inhibition after the period. At ATP concentrations between  $1.35$  and  $40.7 \mu\text{M}$ , the period was  $300 \text{ s}$  and between  $6.8$  to  $40.7 \mu\text{M}$ , the period of  $150 \text{ s}$  was also examined. (B) A Hanes-Woolf type plotting of (A).

ATPase activity as indicated in the figure legends. 50% inhibition of the ATPase reaction was obtained at  $5 \mu\text{M}$  of  $\text{NaN}_3$ . In order to examine the effect of ATP or ADP concentration, we employed  $500 \mu\text{M}$  of  $\text{NaN}_3$ . At a  $\text{NaN}_3$  concentration higher than  $100 \mu\text{M}$ , the rate constant of the inhibition process was essentially independent of the  $\text{NaN}_3$  concentration.

#### Centrifugation elution

Centrifugation elution was carried out according to Penefsky [21] using 1 ml tuberculin syringes and Bio-Gel P-10 (Bio-Rad).

Further experimental details are given in the legends.

## Results

#### Relationship between inhibitory effect of $\text{NaN}_3$ and ATP concentration

Fig. 1 shows typical time-courses of inhibitory action of  $\text{NaN}_3$  on the ATPase activity of SMP. In Fig. 1A, ATPase reaction was started by addition of ATP ( $400 \mu\text{M}$ ) and  $\text{NaN}_3$  ( $500 \mu\text{M}$ ) was added subsequently. The addition of  $\text{NaN}_3$  caused a slow attenuation of ATP

hydrolysis. Final level of attenuation exceeded 95%. This slow attenuation was not an artifact of the response of ATP-regenerating system since oligomycin inhibited the ATPase instantly under the same conditions (data not shown). In Fig. 1B,  $\text{NaN}_3$  was added first and ATP was added subsequently. The results clearly indicate that pre-incubation with  $\text{NaN}_3$  has no effect for inhibition of ATPase activity in SMP and it inhibits the ATPase only after ATP addition. Furthermore, as shown in Fig. 1C, ATPase activity was attenuated more slowly as ATP concentration was decreased ( $8 \mu\text{M}$ ).

As described in the Introduction, the ATPase activity of SMP exhibits negative cooperativity and two apparent  $K_m$ 's of some  $10 \mu\text{M}$  and  $200 \mu\text{M}$  have been calculated [15]. In addition, under single-turnover conditions, a catalytic site with very high affinity has been reported for activated ATPase in SMP [22]. In order to characterize the affinity of the ATP binding which is related to  $\text{NaN}_3$  sensitivity and investigate the relationship between the negative cooperativity observed for ATP hydrolysis, we analyzed the time-course of the inhibition process as follows.

A Guggenheim plot of the time-course of the inhibi-

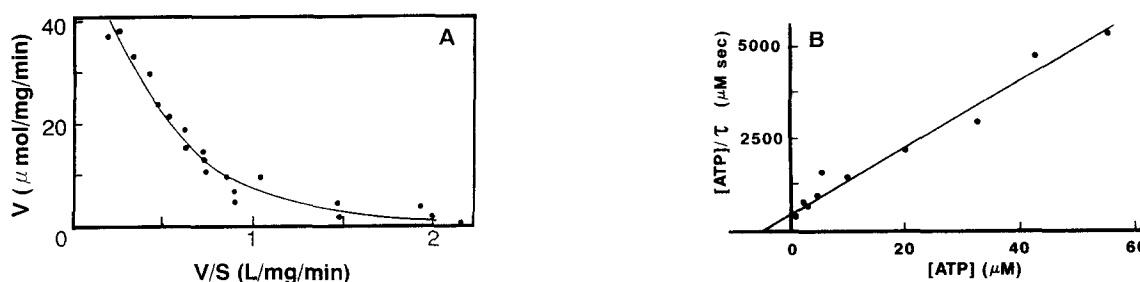


Fig. 3. (A) An Eadie Hofstee plot of ATP hydrolysis by  $F_1$  ATPase. ATP concentration was varied from  $2.5$  to  $2000 \mu\text{M}$ . (B) ATP concentration dependency of the rate constants of  $\text{NaN}_3$  inhibition.  $\text{NaN}_3$  was added after ATP and the rate constants were derived by fitting the time-course of inhibition process with an exponential curve at all the ATP concentrations ( $1.25$  to  $55 \mu\text{M}$ ) examined.

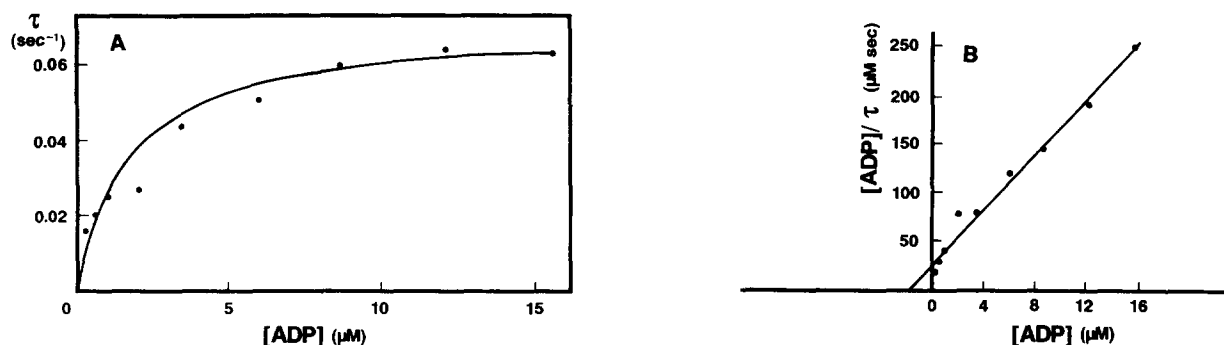


Fig. 4. ADP concentration dependency of the rate constant of inhibition process by  $\text{NaN}_3$ . (A) Rate constants of the inhibition process were plotted against ADP concentrations. SMP were pre-incubated in 50 mM  $\text{TrisSO}_4$ , 2 mM  $\text{MgSO}_4$ , pH 8.0 with various concentrations of ADP and 500 μM of  $\text{NaN}_3$  for 10 to 40 s. Then the remaining ATPase activity was determined by directly adding the aliquots to the ATPase assay mixture containing 400 μM of ATP and 500 μM of  $\text{NaN}_3$ . The ADP present in the pre-incubation medium was quickly converted to ATP on addition to the ATPase assay mixture and had no effect on the measurement. Rate constants of the inhibition process ( $\tau$ ) were derived by fitting the decrease of ATPase activity with an exponential curve taking 100% of activity with the sample pre-incubated without ADP. (B) A Hanes-Woolf type linear plotting of (A).

tion showed that the process was single exponential. On the assumption that the binding of ATP to the site is in rapid equilibrium compared to the inhibition process and the enzyme which binds ATP to the site is inhibited by  $\text{NaN}_3$ , the affinity of the site can be estimated from the rate constant of the inhibition process at various ATP concentration. As shown in Fig. 2A, the plot of rate constant ( $\tau$ ) against ATP concentration gave a hyperbolic curve and a Hanes-Woolf type plotting of Fig. 2A (Fig. 2B) gave a straight line. From the plot, the apparent affinity of the binding site which renders  $\text{NaN}_3$  sensitivity was estimated to be  $14.0 \pm 8.7$  μM ( $\pm$  standard deviation) and  $\tau_{\text{max}}$  was  $0.012 \pm 0.002$  s<sup>-1</sup>. This value was close to the low  $K_m$  of some 10 μM previously reported for ATP hydrolysis [15] and far from  $10^{-12}$  M or 200 μM, strongly suggesting that binding of ATP to the site that represents the apparent  $K_m$  of some 10 μM is responsible for  $\text{NaN}_3$  sensitivity.

The same conclusion was derived as for isolated  $F_1$  ATPase. Our  $F_1$  preparation showed negative cooperativity of ATP hydrolysis characterized by two  $K_m$ 's  $9.0 \pm 8.1$  μM and  $930 \pm 210$  μM (Fig. 3A). Although it was very slowly inhibited by  $\text{NaN}_3$ , even in the absence of adenine nucleotides, the inhibition rate was clearly dependent on ATP concentration and half-maximal promotion of  $\text{NaN}_3$  sensitivity was attained at  $12.6 \pm 7.1$  μM of ATP (Fig. 3B). This is again well consistent with the low  $K_m$  of ATP hydrolysis. However, as the slow inhibition without added adenine nucleotide made some complication in kinetic analyses, we used SMP in the following experiments.

The reversibility of the inhibited state was examined by applying the inhibited SMP to centrifuge columns. It was shown that the presence of only  $\text{NaN}_3$  was

sufficient to maintain the inhibited state, even if ATP was removed\*.

#### Effect of ADP on $\text{NaN}_3$ sensitivity

The fact that inhibitory action of  $\text{NaN}_3$  starts only after ATP addition (Fig. 1B) raised another question whether successive catalytic turnovers are necessary for the inhibition process or not. In order to clarify this point, we pre-incubated SMP with ADP and  $\text{NaN}_3$  in 50 mM  $\text{TrisSO}_4$ , 2 mM  $\text{MgSO}_4$  (pH 8.0) and examined the ATPase activity after centrifuge elution. When SMP were pre-incubated with only ADP (100 μM) for 5 min and passed through a centrifuge column, the ATPase activity was the same as the sample pre-incubated without ADP. When SMP were pre-incubated with ADP and  $\text{NaN}_3$  simultaneously, the ATPase activity was inhibited after passage through a centrifuge column containing  $\text{NaN}_3$ . Like the effect of ATP and  $\text{NaN}_3$ , the inhibited state was maintained by  $\text{NaN}_3$  alone and was reversed when ADP and  $\text{NaN}_3$  were removed. From these results, it can be concluded that successive turnovers of ATPase reaction are not necessary for  $\text{NaN}_3$  inhibition but the presence of bound ADP is enough.

Then, the affinity of the site for ADP was of interest. Pre-incubation of SMP with  $\text{NaN}_3$  and ADP caused time dependent inhibition like that with  $\text{NaN}_3$  and

\* Using [<sup>3</sup>H]ATP, we examined the AT(D)P content of the sample after centrifuge elution to remove AT(D)P in the presence of  $\text{NaN}_3$ . However, although the ATPase activity of the effluent was almost inhibited, we could not detect enough amount of bound label. The use of activated SMP did not improve the data. (Muneyuki, E., and Kamata, H., unpublished results).

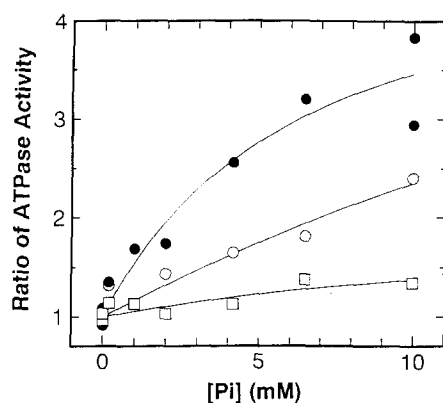


Fig. 5. Effect of inorganic phosphate on  $\text{NaN}_3$  sensitivity. SMP were first incubated with indicated concentration of inorganic phosphate for 1 min and then buffer (□), ADP (13  $\mu\text{M}$ ) (○) or  $\text{NaN}_3$  (500  $\mu\text{M}$ ) and ADP (13  $\mu\text{M}$ ) (●) were added. Just after 40 s, the ATPase activity was measured by directly adding an aliquot to an ATP-regenerating system containing 500  $\mu\text{M}$   $\text{NaN}_3$  and 400  $\mu\text{M}$  ATP. The actual values corresponding to the 1 in ordinate are 0.673, 0.150 and 0.057  $\mu\text{mol mg}^{-1} \text{min}^{-1}$ , for the samples incubated with buffer only (□), ADP (○) and both of ADP and  $\text{NaN}_3$  (●), respectively. Ratios of initial velocity with and without inorganic phosphate are plotted.

ATP. Although the data points are somewhat more scattered than the case of ATP, the rate constants ( $\tau$ ) were derived from the time-course of the inhibition on a similar assumption as in the case of ATP binding. In Fig. 4A, the rate constants ( $\tau$ ) is plotted against ADP concentration. A linear plot was made as Fig. 4B and apparent  $K_d$  for ADP was calculated to be  $1.34 \pm 0.45 \mu\text{M}$ . The corresponding  $\tau_{\text{max}}$  was  $0.066 \pm 0.008 \text{ s}^{-1}$ .

#### Effect of inorganic phosphate on $\text{NaN}_3$ sensitivity

The results obtained above suggests that  $\text{F}_0\text{F}_1$  ATPase in SMP becomes sensitive to  $\text{NaN}_3$  after bound ATP was hydrolyzed to produce bound ADP on the enzyme. Then, it can be expected that addition of inorganic phosphate affects  $\text{NaN}_3$  sensitivity by occupying the  $\gamma$ -phosphate position next to the bound ADP. In order to test this possibility, we examined the effect of inorganic phosphate as shown in Fig. 5. In the absence of inorganic phosphate, pre-incubation with ADP and  $\text{NaN}_3$  caused up to 92% inhibition under the experimental conditions here. On the other hand, inorganic phosphate caused more than 200% activation of the residual activity (Fig. 5A, closed circle). Thus, inorganic phosphate prevents  $\text{NaN}_3$  inhibition. Similar effects were observed for the inhibitory effect of ADP (Fig. 5A, open circle). These effects can be distinguished from the activating oxianion effect of inorganic phosphate, since inorganic phosphate activates ATPase activity only 30% in the absence of ADP or  $\text{NaN}_3$  (Fig. 5A, open squares). Inorganic phosphate may compete with  $\text{NaN}_3$  since at higher concentration of  $\text{NaN}_3$ , the extent of inhibition increased again (data not shown). However, as the relationship between  $\text{NaN}_3$  concentra-

tion and inhibitory effect was not a simple hyperbolic curve, we did not further analyze it quantitatively.

## Discussion

### Nucleotide binding site related to $\text{NaN}_3$ sensitivity

$\text{F}_1$  ATPase has six nucleotide binding sites and three of them are exchangeable, while the other three are nonexchangeable [23]. The three exchangeable sites are thought to be catalytic and apparent  $K_d$  of  $10^{-12} \text{ M}$  under single-turnover condition and apparent  $K_m$  of 1–30  $\mu\text{M}$  and 100–300  $\mu\text{M}$  are thought to correspond to the binding of substrate ATP to each catalytic site [14]. In this study, it was shown that  $\text{F}_0\text{F}_1$  ATPase in submitochondrial particles becomes sensitive to  $\text{NaN}_3$  after it binds ATP or ADP to some nucleotide binding site whose apparent affinity for ATP was  $14.0 \pm 8.7 \mu\text{M}$ . This agreed well with the value previously reported for the lower  $K_m$  of ATPase reaction and proton translocation [15]. The fact that ADP can promote the  $\text{NaN}_3$  sensitivity as ATP means that successive turnovers of the enzyme are not necessary to become inhibited by  $\text{NaN}_3$ . The apparent  $K_d$  for ADP was  $1.34 \pm 0.45 \mu\text{M}$ , which agreed well with the value previously reported by Pedersen [24]. This value may be also compared to the low  $K_m$  for ATP synthesis by activated submitochondrial particles [25,26]. These results indicate that the site of ATP binding related to  $\text{NaN}_3$  is identical with that of ADP binding and the low  $K_m$  mode of ATP hydrolysis corresponds to the low  $K_m$  mode of ATP synthesis.

There have been many reports as to the interaction of  $\text{NaN}_3$  with  $\text{F}_1$  or  $\text{F}_0\text{F}_1$  ATPase [7,16,17,24,27–32]. Noumi et al. have reported that  $\text{NaN}_3$  does not inhibit the uni-site catalysis of  $\text{EF}_1$  [17]. Although the meaning of the uni-site catalysis, particularly as for  $\text{EF}_1$ , is somewhat questionable [33], our present results are consistent with theirs. Recently, Harris has reported the effects of  $\text{NaN}_3$  on mitochondrial  $\text{F}_1$  ATPase [28]. He also observed that at low ATP concentration,  $\text{NaN}_3$  did not inhibit the ATPase activity within 1 min. Although his conclusion is somewhat different from ours, we suspect it might reflect different experimental conditions and direct comparison may be difficult. Our experimental results agree qualitatively well with that obtained by Vasilyeva et al. [16]. They also examined the ATP concentration dependency of the inhibitory effect of  $\text{NaN}_3$  but did not analyze the results in relation to negative cooperativity. In the present study, we have demonstrated that the apparent  $K_d$  of the ATP binding which promotes  $\text{NaN}_3$  inhibition was  $14.0 \pm 8.7 \mu\text{M}$ , well consistent with the low  $K_m$  observed for negative cooperativity of ATP hydrolysis by  $\text{F}_0\text{F}_1$  ATPase of SMP.  $\text{NaN}_3$  sensitivity was also conferred by binding of ADP and the apparent  $K_d$  for ADP agreed with the low  $K_m$  for oxidative phospho-

rylation. The bound ADP was quickly released when subjected to centrifuge elution or dilution in the ATPase assay mixture. At least, the nucleotide binding site characterized here is not a non-exchangeable tight binding site and we prefer a view that the site related to  $\text{NaN}_3$  inhibition is a catalytic site which participates in steady-state catalysis. However, we also recognize it is possible that there may be a rapidly exchangeable regulatory site and our present conclusion is still tentative. As for the chloroplast enzymes, it was concluded that the ADP binding site related to  $\text{NaN}_3$  inhibition was a non-catalytic site [30,34]. As for yeast mitochondrial  $F_1$  ATPase, a mutation in  $\alpha$  subunit which was supposed to be at a non-catalytic site was shown to affect  $\text{NaN}_3$  sensitivity [35].

#### Scheme for $\text{NaN}_3$ inhibition

The present results are summarized in a scheme as shown in Fig. 6. In this, ATP first binds to a catalytic site of enzyme with an affinity of  $10\ \mu\text{M}$  and is hydrolyzed into ADP and inorganic phosphate ( $\text{P}_i$ ). Under steady-state conditions, ADP dissociates quickly from the enzyme before dissociation of  $\text{P}_i$  through the upper pathway of the scheme. On the other hand, to a lesser extent,  $\text{P}_i$  dissociates before ADP and the resultant enzyme  $\cdot$  ADP complex, which is less active, is formed through the lower pathway of the scheme. The enzyme  $\cdot$  ADP complex should not be totally stable and inactive, but slowly dissociates ADP, since the bound ADP can be easily removed by centrifuge elution. The proportion of the enzyme  $\cdot$  ADP complex should be adequately low since the release of  $\text{P}_i$  before ADP is sufficiently slower than that of ADP before  $\text{P}_i$ . However, once formed, the enzyme  $\cdot$  ADP is sensitive to  $\text{NaN}_3$  presumably by forming a ternary enzyme  $\cdot$  ADP  $\cdot$   $\text{NaN}_3$  complex. The scheme presented here is essentially similar to that proposed by Vasilyeva et al. [16]. The effect of  $\text{P}_i$  against  $\text{NaN}_3$  inhibition may be explained as to shift the equilibrium from enzyme  $\cdot$  ADP +  $\text{P}_i$  to  $\text{NaN}_3$  insensitive form of enzyme  $\cdot$  ADP  $\cdot$   $\text{P}_i$ .

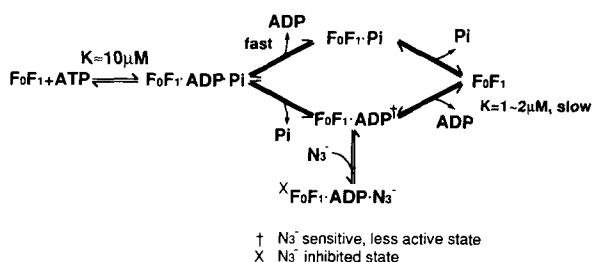


Fig. 6. Scheme for  $\text{NaN}_3$  inhibition on the ATPase activity of submitochondrial particles. At the catalytic site which corresponds to the  $K_m$  of some  $10\ \mu\text{M}$ , the steady-state ATP hydrolysis proceeds mainly through the upper pathway where ADP dissociates quickly before  $\text{P}_i$ . When  $\text{P}_i$  dissociates before ADP, the resultant enzyme  $\cdot$  ADP complex is less active due to slow release of ADP.  $\text{NaN}_3$  acts on this enzyme  $\cdot$  ADP complex. See also the text for details.

$\text{NaN}_3$  has been reported as a unidirectional inhibitor: namely, it inhibits only ATP hydrolysis but not ATP synthesis. This may be because of the presence of  $\text{P}_i$  under ATP synthesis conditions.

Recently, Murataliev et al. have reported that the effect of  $\text{NaN}_3$  is related to Mg inhibition [32]. Although Mg ions are not included in the scheme, all the enzyme  $\cdot$  adenine nucleotide complex should concomitantly contain Mg ions together. The fact that isolated  $F_1$  is inhibited by  $\text{NaN}_3$  even in the absence of ATP or ADP may be explained as that it binds Mg to the site even in the absence of adenine nucleotides. Such enzyme  $\cdot$  metal  $\cdot$  anion complexes have been proved for creatine kinase by infrared spectroscopy [36]. In the case of SMP, the ATPase activity was insensitive to  $\text{NaN}_3$  even in the presence of ADP when Mg was completely removed by EDTA (data not shown), however, the presence of only Mg ions was not sufficient for introduction of  $\text{NaN}_3$  sensitivity as described above.

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